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# PHARMACEUTICS – I

## UNIT 2

### TOPIC :

- **Powders** : Definition, classification, advantages and disadvantages, Simple & compound powders– official preparations, dusting powders, effervescent, efflorescent and hygroscopic powders, eutectic mixtures. Geometric dilutions.



## Powder

→ A powder is a homogeneous mixture of more or less finely divided particle or material in dry form. It is a solid dosage form of medicament which are meant for internal and external uses They are present in crystalline and amorphous form.

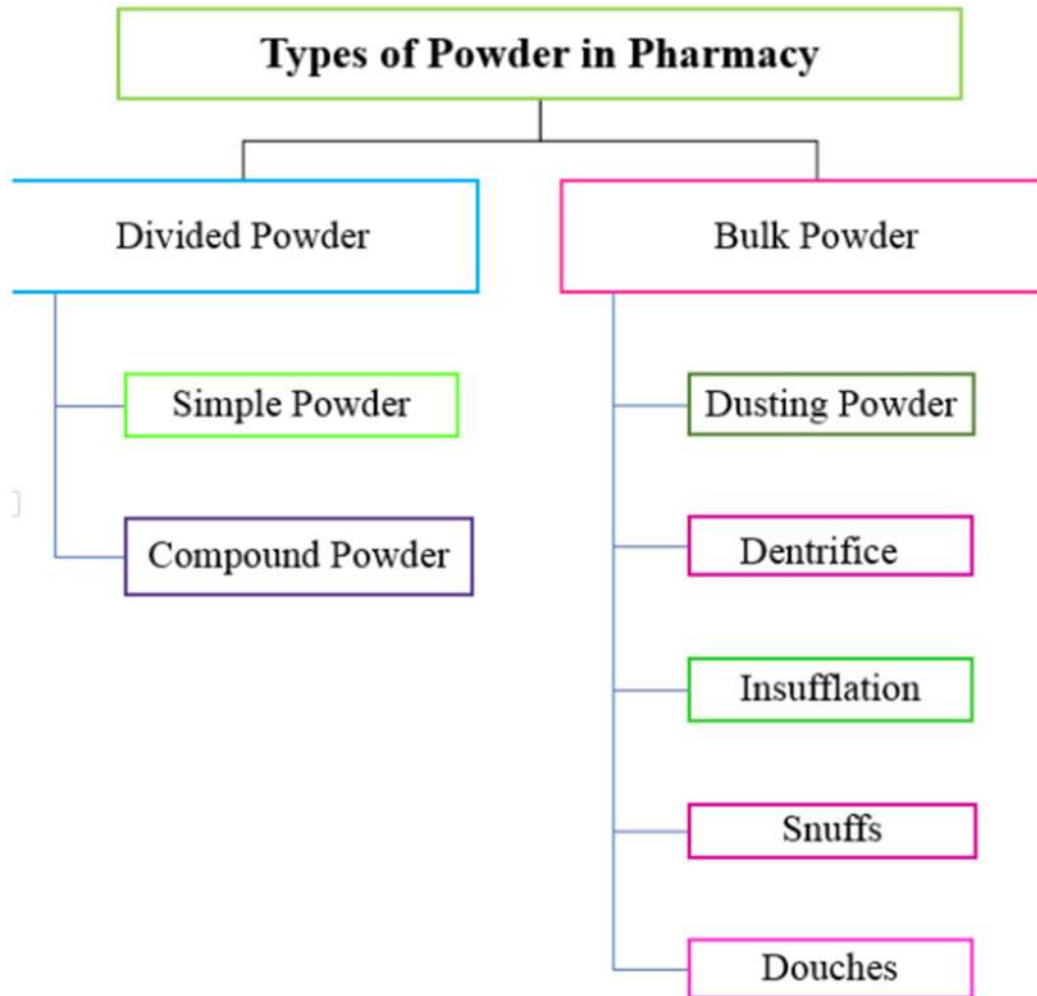
### Advantages

- ✓ They are more stable
- ✓ Easy to handle
- ✓ It is used both internally and externally
- ✓ Chances of incompatibility are less
- ✓ They show rapid therapeutic effect.
- ✓ Ease in administration to all categories of patients.
- ✓ They are economical because they do not require special technique or machinery

### Disadvantages

- ◆ Drugs having bitter nauseous and unpleasant taste cannot be dispensed in powdered form.
- ◆ It gives slow response than liquid dosage form
- ◆ Drugs which get affected by atmospheric conditions are not suitable for dispensing in powder form.
- ◆ Time Consuming Process.

## Classification





## Divided Powder

→ These are unit dose powders normally packed properly

### Simple Powder :

- ◆ These are those type of powder which contain only one ingredient.
- ◆ Which are packed in a folded paper and dispensed in small plastic bags.
- ◆ The powder should not be less than 100 mg
  - Eg : Aspirin

### Compound Powder :

- ◆ Contains two or more than two substances which are mixed together and then divided into individual doses.
  - Eg : Corace SP

## Bulk Powder

- These are those type of powder which are store or dispensed in bulk form I.e. powder are delivered in large quantity
- They are mainly dispense in wide mouth container. It contain non-potent substance like Antacids, Laxatives, etc

## Dusting Powder

- ◆ These are those powder which are meant for external application the skin for local action.
- ◆ They are applied in very fine state to avoid local irritation.
  - Eg : Salicylic acid

### Properties of dusting powder

- Dusting powder should be homogeneous in nature.
- It should have non-irritable property.

- It should be Free flowing.
- Powder should have good spreadability.
- Dusting powder should not be applied to broken skin.

### **Effervescent Powder**

- ◆ They are specially prepared solid dosage form of medicament meant for internal use.
- ◆ They contain medicament (API) mixed with citric acid, tartaric acid and sodium bicarbonate.
- ◆ Saccharine may be added as sweetening agent Before administration the desired quantity is dissolved in water the acid and bicarbonate react together producing effervescence (releasing CO<sub>2</sub>)
- ◆ This mixture should be taken while effervescing
- ◆ Eg : ENO. Aspirin, etc...

### **Dentifrices**

- ◆ These are those substance which are meant for external use and applied with the help of tooth brush for cleaning the surface of teeth.
- ◆ They are available as tooth powder, tooth paste, liquid, and gels.
- ◆ Eg : Saccharine sodium, etc...

### **Efflorescent Powder**

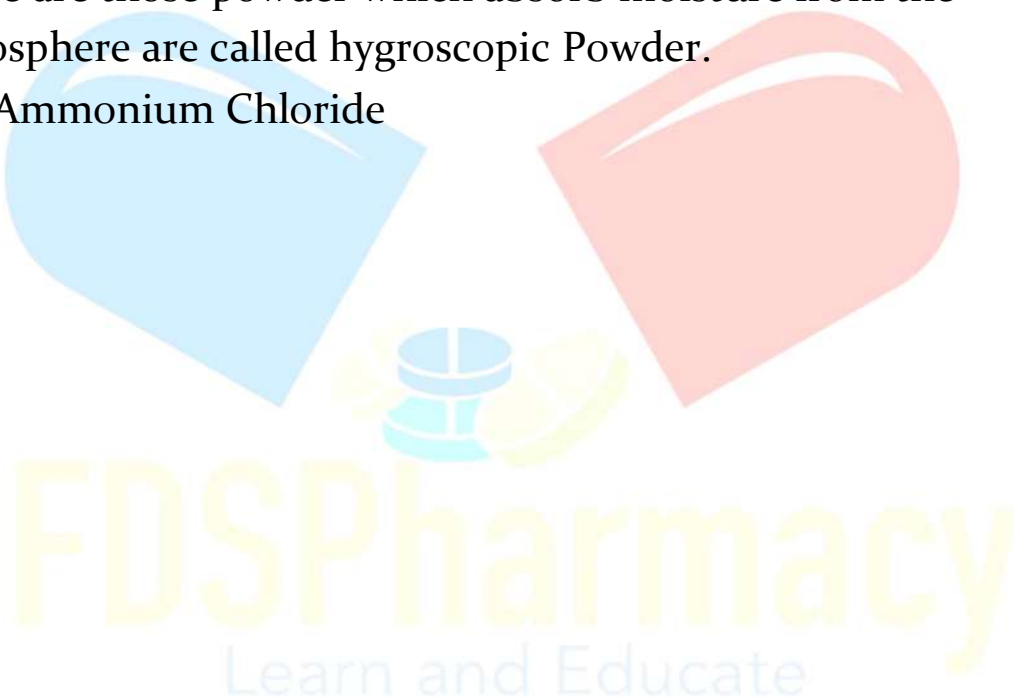
- ◆ When some substances are exposed to air they lose water to the atmosphere there by reducing in weight Solids that behave in this way are those with water of crystallization.
- ◆ The molecules of water of crystallization are partially or completely lost to the atmosphere, thereby making them to lose their crystalline forms



- ◆ Basically, These are those substance which loses water to form a lower hydrate or become anhydrous is termed as efflorescent.
- ◆ Present in form of crystals granules.
- ◆ Eg : Sodium Carbonate, etc...

### **Hygroscopic Powder**

- ◆ These are those powder which absorb moisture from the atmosphere are called hygroscopic Powder.
- ◆ Eg : Ammonium Chloride



## Eutectic Mixtures

- A eutectic mixture is a combination of two or more components that melt or solidify at a single, fixed temperature which is lower than the melting point of any individual component in the mixture. This temperature is called the eutectic point.
- Eg : Salt and Water ( $\text{NaCl} + \text{H}_2\text{O}$ )

## Geometric Dilution

- Geometric dilution (also known as geometric mixing or geometric progression blending) is a method used in pharmacy and chemical mixing to evenly distribute small amounts of a potent substance (like a drug) throughout a larger quantity of diluent or base.
- It ensures uniform distribution of ingredients, especially when the drug quantity is very small compared to the excipient (carrier substance).
- Eg : 15 grams of lactose (diluent)

